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(57) Abstract  This invention provides methods for the treatment of mammal in need thereof an effective amount of a compount of	or inhib and havi	oition ing ac	of ischemia reperfusion injury which cetivity as a leukotriene B <sub>4</sub> antagonist.	omprises administering to a
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## LEUKOTRIENE ANTAGONÁSTS USEFUL FOR TREATING ISCHEMIA REPERFUSION INJURY

During focal ischemia, the extent of parenchymal tissue damage relates closely to two factors: the level of blood flow during the ischemic interval and the duration of ischemia.

On return of blood flow, there is a resumption of the principal functions of tissue perfusion: oxygen delivery, provision of substrates for metabolism, and clearance of metabolic wastes. The restored perfusion counteracts the ischemic injury process and serves to return at least some of the reversibly injured tissue to a functional state.

However, there is a developing consensus that return of blood flow in the postischemic period has a dark side. Interactions between blood and damaged tissue can lead to further tissue injury. It is likely that unmodified blood is not an ideal perfusate for reflow to tissue subjected to prior ischemia. Its reintroduction can actually contribute to parenchymal cell loss as an unwanted side effect of its more familiar capacity to salvage reversibly damaged cells.

This paradoxically harmful aspect of blood flow return has been termed reperfusion injury and the process has potential relevance to clinical medicine. Although reperfusion per se is essential to functional recovery and reduces the volume of tissue infarcted, a further reduction in tissue loss might be achieved by measures that counteract reperfusion injury.

Virtually every organ undergoing ischemic insult may be subjected to reperfusion injury, although much of the work on these phenomena has been done in heart and brain.

Concepts of reperfusion injury tend to emphasize similarities between mechanisms of cell death in various tissues rather than properties unique to a given tissue.

Early research on myocardial reperfusion suggested that reperfusion may accelerate the development of necrosis in

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irreversibly injured myocytes. An ultrastructural appearance of "explosive swelling," was noted which included architectural disruption, contraction bands, and intermitochondrial calcium phosphate granules.

Later studies reported the paradox of myocardial necrosis after successful revascularization by coronary artery bypass graft surgery and suggested that the lesions were operation related and represented contracture due to calcium loading and myocardial cellular edema in the distribution of widely patent arterial grafts. The studies further concluded that prevention of intraoperative myocardial injury must also focus on characteristics of the phase of myocardial reperfusion.

Additional demonstrated that measures instituted after the termination of ischemia, which attenuated the rise in myocardial cytosolic calcium, led to a reduction of tissue injury. Subsequently, other workers have implicated cell swelling, white blood cell plugging of vessels, and free radical damage in reperfusion injury of the heart.

Research also focused attention on the contribution of reperfusion impairment to neural injury. Based on a disproportionately long survival of retinal ganglion cells exposed to anoxia in culture (in which oxygen could diffuse directly to the cells when reintroduced rather than requiring an intact tissue perfusion) as compared with the rapidity with which ischemia in vivo leads to irreversible brain cell damage, they postulated that recirculation of ischemic brain could become compromised and account for the differences in neuronal vulnerability observed under the two conditions.

This impairment of microvascular reperfusion was attributed primarily to two phenomena. The first was narrowing of the capillary lumen due to perivascular swelling and formation of endothelial blebs, and the second was increased blood viscosity. Fluid shifts into cells, with their sodium-potassium adenosine triphosphatase pumps shut down by energy failure, were thought to underlie both

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the capillary narrowing and the microvascular hemoconcentration with increased blood viscosity. In addition, others described the topography of "no reflow," and published provocative studies that have demonstrated the resumption of function in cortical neurons after total ischemia for periods of 30 minutes to 1 hour if measures were taken to overcome reperfusion impairment.

The concept of a multifocal impairment of reflow manifesting immediately after ischemia and decreasing with time was modified and extended by consideration of "autodestruction" in models of spinal cord trauma and reperfusion problems encountered during autotransplantation of organs preserved by ex vivo perfusion.

Returning blood flow was viewed as having two dichotomous effects: (1) a well-established restorative effect and (2) a postulated capacity to undergo a multifactoral interaction with damaged tissue that could progressively shut down microcirculatory flows and contribute to further neuronal damage.

Studies measuring blood flow and metabolism in gerbils subjected to unilateral carotid occlusion, noted an immediate return of perfusion to the brain on release of the occlusion, following in 10 to 30 minutes by a decline in flow for at least 4 hours with a coincident rise in the cerebral metabolic rate that disrupted the normal flow-metabolism couple. This phenomenon was later documented in the four-vessel occlusion model that they had devised in rats.

Consideration of blood elements, blood vessels, and blood flow during and after ischemia is also derived from work in endothelial cell tissue culture. Recent evidence indicates that endothelial cells produce superoxide under certain circumstances. This may have a major role in reperfusion injury.

Calcium has been implicated in reperfusion injury under several circumstances. Reperfusion can cause an acute acceleration of calcium influx into cells by injuring cell

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membranes with a consequent increased membrane permeability. The exact role of cytosolic calcium accumulation in the evolution of neuronal and myocardial injury after ischemia remains to be elucidated, but there is considerable evidence that ischemia and reperfusion are associated with a severe disruption of calcium homeostasis.

Generally, reperfusion of ischemic tissues is associated with endothelial cell injury, inflammatory cell infiltration and microvascular dysfunction. "Reperfusion injury" has been attributed to leukocytes and their ability to increase free radicals, cytotoxic enzymes and affect blood rheology. However, providing blood soon to ischemic tissue is, of course, beneficial.

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Research in the area of allergic reactions of the lung has provided evidence that arachidonic acid derivatives formed by the action of lipoxygenases are related to various disease states. Some of these arachidonic acid metabolites have been classified as members of a family of eicosatetraenoic acids termed leukotrienes. Three of these substances are currently thought to be major components of what has been previously called slow reacting substance of anaphylaxis (SRS-A) and have been designated leukotrienes C4, D4, and E4 (LTC4, LTD4, and LTE4, respectively).

Another arachidonic acid metabolite, leukotriene B4 (LTB4), is a proinflammatory lipid which has been implicated in the pathogenesis of psoriasis, arthritis, chronic lung diseases, acute respiratory distress syndrome, shock, asthma, inflammatory bowel diseases, and other inflammatory states characterized by the infiltration and activation of polymorphonuclear leukocytes and other proinflammatory cells. Thus activated, the polymorphonuclear leukocytes liberate tissue-degrading enzymes and reactive chemicals causing the inflammation. Antagonism of LTB4 should therefore provide a novel therapeutic approach to treatment of these and other LTB4 mediated conditions.

Because of the debilitating effects of ischemia reperfusion injury, there continues to exist a need for effective treatments.

This invention provides a method for the treatment or inhibition of ischemia reperfusion injury in a mammal comprising administering to a mammal in need thereof an effective amount of a compound of the Formula I

$$R_{2}$$
 $R_{2}$ 
 $R_{1}$ 
 $R_{1}$ 

wherein:

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R<sub>1</sub> is C<sub>1</sub>-C<sub>5</sub> alkyl, C<sub>2</sub>-C<sub>5</sub> alkenyl, C<sub>2</sub>-C<sub>5</sub> alkynyl, C<sub>1</sub>-C<sub>4</sub> alkoxy, (C<sub>1</sub>-C<sub>4</sub> alkyl)thio, halo, or R<sub>2</sub>-substituted phenyl;

each R<sub>2</sub> and R<sub>3</sub> are each independently hydrogen, halo, hydroxy, C<sub>1</sub>-C<sub>4</sub> alkyl, C<sub>1</sub>-C<sub>4</sub> alkoxy, (C<sub>1</sub>-C<sub>4</sub> alkyl)-S(0)<sub>q</sub>-, trifluoromethyl, or di-(C<sub>1</sub>-C<sub>3</sub> alkyl) amino;

X is -O-, -S-, -C(=O), or  $-CH_2-$ ;

25 Y is -O- or -CH<sub>2</sub>-;

or when taken together, -X-Y- is -CH=CH- or

—c=c ;

Z is a straight or branched chain C1-C10 alkylidenyl;

A is a bond, -O-, -S-, -CH=CH-, or -CRaRb-, where  $R_{a}$  and  $R_{b}$  are each independently hydrogen, C1-C5 alkyl, or R7-substituted phenyl, or when taken together with the carbon atom to which they are attached form a C4-C8 cycloalkyl ring;

R<sub>4</sub> is R<sub>6</sub>

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$$R_{11}$$

$$W-R_6$$
 or  $R_7$   $W-R_6$ 

where,

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each R6 is independently -COOH, 5-tetrazoly1,
-CON(R9)2, or -CONHSO2R10;

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each R7 is hydrogen, C1-C4 alkyl, C2-C5 alkenyl, C2-C5 alkynyl, benzyl, methoxy, -W-R6, -T-G-R6, (C1-C4 alkyl)-T-(C1-C4 alkylidenyl)-O-, or hydroxy;

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Rg is hydrogen or halo;

each R9 is independently hydrogen, phenyl, or C1-C4 alkyl, or when taken together with the

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nitrogen atom form a morpholino, piperidino, piperazino, or pyrrolidino group;

R<sub>10</sub> is C<sub>1</sub>-C<sub>4</sub> alkyl or phenyl;

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 $R_{11}$  is  $R_2$ , -W-R6, or -T-G-R6;

each W is a bond or straight or branched chain divalent hydrocarbyl radical of one to eight carbon atoms;

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each G is a straight or branched chain divalent hydrocarbyl radical of one to eight carbon atoms;

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each T is a bond, -CH2-, -O-, -NH-, -NHCO-, -C(=O)-, or -S(O) $_{\mathbf{q}}$ -;

K is -C(=0) - or -CH(OH) -;

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each q is independently 0, 1, or 2;

p is 0 or 1; and

t is 0 or 1;

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provided when X is -O- or -S-, Y is not -O-;

provided when A is -O- or -S-, R4 is not R6;

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provided when A is -O- or -S- and Z is a bond, Y is not -O-; and

provided W is not a bond when p is 0;

35 or a pharmaceutically acceptable salt or solvate thereof.

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The following definitions refer to the various terms used throughout this disclosure.

The term "C1-C5 alkyl" refers to the straight and branched aliphatic radicals of 1 to 5 carbon atoms such as methyl, ethyl, propyl, isopropyl, n-butyl, sec-butyl, tert-butyl, n-pentyl, 2,2-dimethylpropyl, and the like. Included within this definition are the terms "C1-C3 alkyl", and "C1-C4 alkyl".

The term "C2-C5 alkenyl" refers to straight and branched aliphatic radicals of 2 to 5 carbon atoms containing one double bond, such as -CH=CH2, -CH2CH=CH2, -CH2CH=CH2, -CH2C(CH3)=CH2, -CH2CH=C(CH3)2, and the like.

The term "C<sub>2</sub>-C<sub>5</sub> alkynyl" refers to straight and branched aliphatic residues of 2 to 5 carbon atoms containing one triple bond, such as  $-C \equiv CH$ ,  $-CH_2-C \equiv CH$ ,  $-CH_2-C \equiv CH$ ,  $-CH_2-C \equiv CH$ ,  $-CH_2-C \equiv CH$ , and the like.

The term "C1-C4 alkoxy" refers to methoxy, ethoxy, propoxy, isopropoxy, butoxy, sec-butoxy, and tert-butoxy.

The term "halo" refers to fluoro, chloro, bromo, and iodo.

The term " $C_1$ - $C_{10}$  alkylidenyl" refers to a divalent radical derived from a  $C_1$ - $C_{10}$  alkane such as - $CH_2$ -,

 $-CH(CH_3)$  -,  $-C(CH_3)_2$ -,  $-CH(C_2H_5)$ -,  $-CH_2CH_2$ -,  $-CH_2CH(CH_3)$ -,

 $-CH(CH_3)CH_2-$ ,  $-CH(CH_3)CH(CH_3)-$ ,  $-CH_2C(CH_3)_2-$ ,  $-CH_2CH(C_2H_5)-$ ,

25  $-CH_2CH_2CH_2-$ ,  $-CH(CH_3)CH_2CH_2-$ ,  $-CH_2CH(CH_3)CH_2-$ ,

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 $-CH_2CH(C_2H_5)CH_2-$ ,  $-CH_2CH_2CH(C_2H_5)-$ ,  $-C(CH_3)_2CH_2CH_2-$ ,

-CH(CH<sub>3</sub>)CH<sub>2</sub>CH(CH<sub>3</sub>)-, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-,

 $-CH_2C(CH_3)_2CH_2-$ ,  $-CH_2CH_2CH(C_2H_5)CH_2-$ ,  $-CH_2CH_2CH_2CH_2CH_2-$ ,

-CH(CH<sub>3</sub>)CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>-, -(CH<sub>2</sub>)<sub>10</sub>-, and the

like. Included within this definition are the terms " $C_1$ - $C_4$  alkylidene" and " $C_2$ - $C_4$  alkylidene".

The term "C4-C8 cycloalkyl" refers to a cycloalkyl ring of four to eight carbon atoms, such as cyclobutyl, cyclopentyl, cyclohexyl, 4,4-dimethylcyclohexyl, cycloheptyl, cyclooctyl, and the like.

The term "straight or branched chain divalent hydrocarbyl residue of one to eight carbon atoms" refers to

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a divalent radical derived from a straight or branched alkane, alkene, or alkyne of one to eight carbon atoms. Depending upon the branching and number of carbon atoms, as will be appreciated by organic chemists, such a moiety can contain one, two or three double or triple bonds, or combinations of both. As such, this term can be considered an alkylidene group as defined above containing from 1 to 8 carbon atoms optionally containing one to three double or triple bonds, or combinations of the two, limited as noted in the preceding sentence.

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This invention includes the pharmaceutically acceptable base addition salts of the compounds of Formula I. Such salts include those derived from inorganic bases, such as ammonium and alkali and alkaline earth metal hydroxides, carbonates, bicarbonates, and the like, as well as salts derived from basic organic amines, such as aliphatic and aromatic amines, aliphatic diamines, hydroxy alkylamines, and the like. Such bases useful in preparing the salts of this invention thus include ammonium hydroxide, potassium carbonate, sodium bicarbonate, calcium hydroxide, methyl amine, diethyl amine, ethylene diamine, cyclohexylamine, ethanolamine, and the like. The potassium and sodium salt forms are particularly preferred.

This invention includes both mono-salt forms, ie, a 1:1 ratio of a compound of Formula I with a base as previously described, as well as di-salt forms in those instances where a compound of Formula I has two acidic groups. In addition, this invention includes any solvate forms of the compounds of Formula I or salts thereof, such as ethanol solvates, hydrates, and the like.

It is recognized that in compounds having branched alkyl, alkylidenyl, or hydrocarbyl functionality, and in those compounds bearing double or triple bonds, various stereoisomeric products may exist. This invention is not limited to any particular stereoisomer but includes all possible individual isomers and mixtures thereof. The term

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"5-tetrazolyl" refers to both tautomers, ie, (1H)-5-tetrazolyl and (2H)-5-tetrazolyl.

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A most preferred group of compounds employed in the methods of the present invention are those compounds of Formula Ia:

$$R_2$$
 O-CH<sub>2</sub>-Z-A-R<sub>1</sub>

and pharmaceutically acceptable base addition salts thereof. Especially preferred are those compounds wherein R2 is halo, particularly fluoro. Preferred R1 substituents are propyl and especially ethyl.

Preferred Z substituents include  $C_2-C_4$  alkylidene, particularly  $-CH_2CH_2-$  and  $-CH_2CH_2CH_2CH_2-$ . Preferred A groups include -O-,  $-CH_2-$ ,  $-CH(R_7-substituted phenyl)-$ , and  $-C(CH_3)_2$ .

Preferred R<sub>4</sub> groups include -COOH, 5-tetrazolyl, or a mono-, di-, or tri-cyclic group as drawn above wherein there is at least one acidic group attached to a ring, such as -W-COOH, -T-G-COOH, or the corresponding tetrazole derivatives. The preferred W moiety is that of a bond or straight chain C<sub>1</sub>-C<sub>4</sub> alkylidene; preferred G moieties are straight chain C<sub>1</sub>-C<sub>4</sub> alkylidene. It is preferred that R<sub>5</sub> or R<sub>7</sub> be C<sub>1</sub>-C<sub>4</sub> alkyl, especially n-propyl.

Particularly preferred groups are those wherein A is  $-CH(R_7-substituted\ phenyl)-$  and R4 is -COOH or 5-tetrazolyl. Also preferred are those compounds wherein A is -O- and R4 is

Preferred aspects of this substructure are those therein  $R_7$  is  $C_1-C_4$  alkyl, especially n-propyl, and  $R_6$  is -W-COOH. Particularly preferred are those compounds wherein T is -O- or -S- and W is a bond.

Particularly preferred compounds of the instant invention include 2-[2-propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]phenoxylbenzoic acid:

10 3-(2-(3-(2-ethyl-4-(4-fluorophenyl)-5hydroxyphenoxy)propoxy)-6-(4-carboxyphenoxy)phenyl)propionic acid; 1-(4-(carboxymethoxy)phenyl)-1-(1H-tetrazol-5-yl)-6-(2-ethyl-4-(4fluorophenyl)-5-hydroxyphenoxy)hexane; 3-[4-[7-carboxy-9oxo-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]-9H-xanthene]]propanoic acid; 5-[3-[2-(1-carboxy)ethyl]-4-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]phenyl]-4-pentynoic acid or a pharmaceutically
acceptable salt or solvate thereof.

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The leukotriene B<sub>4</sub> (LTB<sub>4</sub>) antagonists employed in the methods of the present invention may be synthesized essentially as described in US Patent No. 5,462,954 issued October 31, 1995, the entire contents of which are herein incorporated by reference.

The following examples further illustrate the preparation of the intermediates and compounds employed in this invention. The examples are illustrative only and are not intended to limit the scope of the invention. Melting points were determined on a Thomas-Hoover apparatus and are uncorrected. NMR spectra were determined on a GE QE-300 spectrometer. All chemical shifts are reported in parts per million (\_) relative to tetramethylsilane. Chemical shifts

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of aromatic protons of quinoline species in DMSO-d6 are concentration dependent. The following abbreviations are used to denote signal patterns: s = singlet, d = doublet, t = triplet, q = quartet, b = broad, m = multiplet. Infrared spectra were determined on a Nicolet DX10 FT-IR spectrometer. Mass spectral data were determined on a CEC-21-110 spectrometer using electron impact (EI) conditions, a MAT-731 spectrometer using free desorption (FD) conditions, or a VG ZAB-3F spectrometer using fast atom bombardment (FAB) conditions. Silica gel chromatography was performed 10 using ethyl acetate/hexane gradients unless otherwise indicated. Reverse-phase chromatography was performed on MCI CHP20P gel using an acetonitrile/water or methanol/watergradient unless otherwise indicated. Tetrahydrofuran (THF) was distilled from sodium/benzophenone ketyl immediately 15 prior to use. All reactions were conducted under argon atmosphere with stirring unless otherwise noted. structures were confirmed by infra-red, proton nuclear magnetic resonance, or mass spectral analysis, the compound is so designated by "IR", "NMR", or "MS", respectively.

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#### Example 1

3-[2-[3-[(5-Ethyl-2-hydroxy[1,1'-biphenyl]-4yl)oxy]propoxy]-1-dibenzofuran]propanoic acid disodium salt

HO COOEL

ONA

COONA

COONA

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A. Preparation of 3,3-diethoxy-2,3-dihydro-1H-benzofuro-[3,2-f][1]benzopyran.

A solution of 2-hydroxydibenzofuran (5.00 g, 27.2 mmol), triethylorthoacrylate (10.1 g, 54.3 mmol) and pivalic acid (1.39 g, 13.6 mmol) in toluene (100 mL) was refluxed for 18 hours. The mixture was cooled to room temperature and washed once with water and once with a saturated sodium bicarbonate solution, dried over sodium sulfate, filtered and concentrated in vacuo to provide an orange oil. This material was diluted with hexane and maintained at  $-20^{\circ}$ C for 18 hours. The resulting crystals were collected via vacuum filtration to provide 5.67 g (67%) of the desired title intermediate, mp 64°C; NMR (CDCl<sub>3</sub>) 7.96 (d, J = 7.8 Hz, 1H),

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7.57 (d, J = 8.0 Hz, 1H), 7.46 (t, J = 8 Hz, 1H), 7.35 (m, 2H), 7.06 (d, J = 8.8 Hz, 1H), 3.82 (q, J = 7.2 Hz, 2H), 3.73 (q, J = 6.8 Hz, 2H), 3.35 (t, J = 6.9 Hz, 2H), 2.29 (t, J = 7.0 Hz, 2H), 1.23 (t, J = 7.1 Hz, 6H); MS-FD m/e 312 (p); IR (CHCl<sub>3</sub>, cm<sup>-1</sup>) 2982, 1494, 1476, 1451, 1434, 1251, 1090, 1054, 975.

Analysis for C<sub>19</sub>H<sub>20</sub>O<sub>4</sub>:

Calc: C, 73.06; H, 6.45; Found: C, 72.81; H, 6.72.

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B. Preparation of 3-[1-(2-hydroxydibenzofuran)]propanoic acid ethyl ester.

A mixture of 3,3-diethoxy-2,3-dihydro-1H-benzofuro-[3,2f][1]benzopyran (3.50 g, 11.2 mmol) and 10% aqueous 15 hydrochloric acid (5 mL) in ethyl acetate (30 mL) was stirred at room temperature for 1 hour. The resulting mixture was washed once with water, dried over sodium sulfate, filtered and concentrated in vacuo to provide a tan solid. Recrystallization from hexane/ethyl acetate provided 20 3.11 g (98%) of the desired title intermediate as an offwhite crystalline material: mp 128-131°C; NMR (CDCl<sub>3</sub>) 7.88 (d, J = 7.7 Hz, 1H), 7.59 (d, J = 8.4 Hz, 1H), 7.47 (t, J =7.2 Hz, 1H), 7.37 (d, J = 8.9 Hz, 1H), 7.36 (t, J = 6.6 Hz, 1H), 7.13 (d, J = 8.8 Hz, 1H), 7.13 (q, J = 8.8 Hz, 2H), 25 3.43 (t, J = 5.8 Hz, 2H), 3.01 (t, J = 7.7 Hz, 2H), 1.23 (t, J = 7.2 Hz, 3H); MS-FD m/e 284 (100, p), 256 (65), 238 (17);IR  $(KBr, cm^{-1})$  2985 (b), 1701, 1430, 1226, 1183, 1080. Analysis for  $C_{17}H_{16}O_4$ :

30 Calc: C, 71.82; H, 5.67; Found: C, 71.90; H, 5.43.

C. Preparation of 3-[2-[3-[[5-ethyl-2-(phenylmethoxy)-[1,1'-biphenyl]-4-yl]oxy]propoxy]-1-dibenzofuran]propanoic acid ethyl ester.

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3-[1-(2-Hydroxydibenzofuran)]propanoic acid ethyl ester (625 mg, 2.20 mmol) was dissolved in dimethylformamide (10 mL) and carefully treated at room temperature with 95% sodium hydride (58 mg, 2.4 mmol). When gas evolution had ceased, 2-benzyloxy-1-phenyl-5-ethyl-4-(3-chloro-1-propyloxy)benzene (836 mg, 2.20 mmol) was added and the resulting mixture was stirred for 18 hours. The mixture was diluted with ether and washed once with water. The organic layer was dried over sodium sulfate, filtered, and concentrated in vacuo to provide a dark oil. Silica gel chromatography (ethyl acetate/hexane) provided 200 mg (14%) of the desired titled intermediate as a colorless oil: NMR (CDCl<sub>3</sub>) 8.11 (d. J = 7.7 Hz, 1H), 7.57 (m. 3H), 7.48 (t. J =

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7.3 Hz, 1H), 7.20-7.44 (m, 10 H), 7.17 (s, 1H), 7.08 (d, J = 8.9 Hz, 1H), 6.67 (s, 1H), 5.05 (s, 2H), 4.29 (t, J = 6.2 Hz, 2H), 4.26 (t, J = 6.1 Hz, 2H), 4.15 (q, J = 7.2 Hz, 2H), 3.54 (t, J = 8.5 Hz, 2H), 2.67 (m, 4H), 2.37 (t, J = 6.0 Hz, 2H), 1.21 (m, 6H).

20 D. Preparation of 3-[2-[3-[(5-ethyl-2-hydroxy[1,1'-biphenyl]-4-yl)oxy]propoxy]-1-dibenzofuran]propanoic acid disodium salt.

To a nitrogen-purged solution of 3-[2-[3-[[5-ethyl-2-(phenylmethoxy)[1,1'-biphenyl]-4-yl]oxy)propoxy]-1-25 dibenzofuran]propanoic acid ethyl ester (200 mg, 0.318 mmol) in a 1:1 mixture of methanol/tetrahydrofuran (40 mL) was added 10% palladium on carbon (25 mg). The resulting suspension was hydrogenated at 1 atm pressure for 24 hours at room temperature. The mixture was filtered through a 30 short pad of Florisil® and the filtrate concentrated in vacuo. The residue was dissolved in a 1:1 mixture of methanol/tetrahydrofuran (20 mL) and treated with 5N sodium hydroxide solution (2 mL) at room temperature for 24 hours. The resulting mixture was extracted once with diethyl ether. 35 The aqueous layer was acidified with 5N hydrochloric acid solution and extracted twice with methylene chloride. The

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combined methylene chloride fractions were concentrated in vacuo. The residue was dissolved in a minimum of 1N sodium hydroxide solution and purified on HP-20 resin to provide 53 mg (30%) of the desired title product as a fluffy white solid: NMR (DMSO-d<sub>6</sub>) 8.12 (d, J = 6.9 Hz, 1H), 7.64 (d, J = 8.2 Hz, 1H), 7.37-7.57 (m, 5H), 7.30 (m, 2H), 7.14 (m, 2H), 6.96 (s, 1H), 6.93 (s, 1H), 4.30 (t, J = 7.3 Hz, 2H), 4.14 (t, J = 5.4 Hz, 2H), 2.48 (m, 4H), 2.23 (m, 4H), 1.10 (t, J = 7.6 Hz, 3H); MS-FAB m/e 555 (88, p + 1), 533 (62); IR (CHCl<sub>3</sub>, cm<sup>-1</sup>) 3384 (b), 2969, 1566, 1428, 1257, 1181.

Analysis for C32H28O6Na2:

Calc: C, 69.31; H, 5.09; Found: C, 69.51; H, 5.39.

15 Example 2

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7-Carboxy-9-oxo-3-[3-(2-ethyl-5-hydroxy-4-phenylphenoxy)propoxy]-9H-xanthene-4-propanoic acid disodium salt monohydrate

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A mixture of 2-benzyloxy-1-phenyl-5-ethyl-4-(3-chloro-1-propyloxy)benzene (749 mg, 1.97 mmol), ethyl 7-carboethoxy-3-hydroxy-9-oxo-9H-xanthene-4-propanoate (729 mg, 1.97 mmol), potassium carbonate (1.36 g, 9.85 mmol) and potassium iodide (33 mg, 0.20 mmol) was refluxed for 24 hours. Dimethylsulfoxide (2 mL) was added and heating continued for 24 hours. The reaction mixture was cooled to room temperature, diluted with ethyl acetate, and washed once with water. The organic layer was dried over sodium

-18-

sulfate, filtered and concentrated in vacuo to reveal a tan This material was dissolved in ethyl acetate (30 mL) and the resulting solution purged with nitrogen. To this solution was added 10% palladium on carbon (120 mg) and the resulting suspension hydrogenated at 1 atmosphere of pressure. The solution was filtered and concentrated in vacuo to provide a colorless oil. This material was dissolved in a solution of 1:1 methanol/tetrahydrofuran (30 mL) and treated with 5N sodium hydroxide solution (2 mL) at room temperature for 18 hours. The resulting solution was 10 extracted once with diethyl ether and the aqueous layer acidified with 5N hydrochloric acid solution. The resulting precipitate was collected via suction filtration. This material was converted to the di-sodium salt and purified as described above for the preparation of Example 1(D) to 15 provide 390 mg (56%) of the desired title product as a fluffy white solid: NMR (DMSO- $d_6$ ) 12.65 (s, 1H, -OH), 8.65 (s, 1H), 8.28 (dd, J = 8.5, 2.0 Hz, 1H), 8.01 (d, J = 8.9)Hz, 1H), 7.50 (m, 3H), 7.29 (t, J = 7.8 Hz, 2H), 7.17 (m, 2H), 6.93 (s, 1H), 6.89 (s, 1H), 4.26 (m, 4H), 3.12 (m, 2H), 20 2.47 (m, 2H), 2.23 (m, 2H), 1.10 (t, J = 7.4 Hz, 3H); MS-FAB m/e 627 (24, p), 605 (40), 583 (24), 331 (24), 309 (100); IR  $(KBr, cm^{-1})$  3419 (b), 2962, 1612, 1558, 1443, 1390, 1277, 1084.

Analysis for C34H28O9Na2·H2O:

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Calc: C, 63.34; H, 4.69;

Found: C, 63.36; H, 4.50.

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#### Example 3

2-[2-Propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]phenoxy]benzoic acid sodium salt

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A. Preparation of 2-[2-propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-(phenylmethoxy)phenoxy]propoxy]phenoxy]
10 benzoic acid methyl ester.

A mixture of 2-benzyloxy-1-(4-fluorophenyl)-5-ethyl-4-(3-chloro-1-propyloxy)benzene (20.0 g, 50.2 mmol) and sodium

iodide (75.3 g, 502 mmol) in 2-butanone (200 mL) was refluxed for 6 hours. The mixture was diluted with ether and washed once with water. The organic layer was dried over sodium sulfate, filtered, and concentrated in vacuo to provide a colorless oil. This material was dissolved in dimethylformamide (100 mL) and treated with 2-(3-hydroxy-2propylphenoxy)benzoic acid methyl ester (14.4 g, 50.2 mmol) and potassium carbonate (20.8 g, 151 mmol) at room temperature for 24 hours. This mixture was diluted with water and twice extracted with ether. The aqueous layer was 10 separated and back-extracted once with ethyl acetate. The combined organic layers were dried over sodium sulfate, filtered, and concentrated in vacuo to provide a yellow oil. Silica gel chromatography provided 25.4 g (78%) of the desired title intermediate as a pale golden oil: NMR 15  $(CDCl_3)$  7.91 (d, J = 7.8 Hz, 1H), 7.54 (d, J = 8.6 Hz, 1H), 7.52 (d, J = 8.5 Hz, 1H), 7.25-7.43 (m, 6H), 7.03-7.38 (m, 5H), 6.84 (d, J = 8.3 Hz, 1H), 6.71 (d, J = 8.1 Hz, 1H), 6.63 (s, 1H), 6.47 (d, J = 8.1 Hz, 1H), 5.03 (s, 2H), 4.24(t, J = 5.7 Hz, 2H), 4.21 (t, J = 5.8 Hz, 2H), 3.86 (s, 3H),20 2.69 (t, J = 7.8 Hz, 2H), 2.64 (t, J = 7.7 Hz, 2H), 2.34(quintet, J = 6.0 Hz, 2H), 1.60 (hextet, J = 5.0 Hz, 2H), 1.22 (t, J = 7.5 Hz, 3H), 0.94 (t, J = 7.5 Hz, 3H); MS-FD m/e 648 (p); IR (CHCl<sub>3</sub>, cm<sup>-1</sup>) 2960, 1740, 1604, 1497, 1461, 25 1112.

Analysis for C41H41O6F:

Calc: C, 75.91; H, 6.37; Found: C, 76.15; H, 6.45.

- 30 B. Preparation of 2-[2-propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]phenoxy]benzoic acid methyl ester.
- 2-[2-Propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-35 (phenylmethoxy)phenoxy]propoxy]phenoxy]benzoic acid methyl ester (33.0 g, 50.9 mmol) was de-benzylated as described

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above for the preparation of Example 2 to provide 27.3 g (96%) of the title intermediate as an amber oil: NMR (CDC1<sub>3</sub>) 7.90 (dd, J = 7.8, 1.7 Hz, 1H), 7.42 (m, 3H), 7.05-7.23 (m, 4H), 6.99 (s, 1H), 6.84 (d, J = 8.1 Hz, 1H), 6.70 (d, J = 8.1 Hz, 1H), 6.55 (s, 1H), 6.46 (d, J = 8.1 Hz, 1H), 5.05 (s, 1H, -OH), 4.23 (m, 4H), 3.86 (s, 3H), 2.68 (t, J = 7.4 Hz, 2H), 2.62 (q, J = 7.5 Hz, 2H), 2.36 (quintet, J = 6.0 Hz, 2H), 1.60 (hextet, J = 7.7 Hz, 2H), 1.20 (t, J = 7.6 Hz, 3H), 0.94 (t, J = 7.4 Hz, 3H); MS-FD m/e 558 (p); IR (CHCl<sub>3</sub>, cm<sup>-1</sup>) 2965, 1727, 1603, 1496, 1458, 1306, 1112.

Analysis for C34H35O6F:

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Calc: C, 73.10; H, 6.31; Found: C, 73.17; H, 6.42.

15 C. Preparation of 2-[2-propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]phenoxy]benzoic acid sodium salt.

2-[2-Propy1-3-[3-[2-ethy1-4-(4-fluoropheny1)-5hydroxyphenoxy]phenoxy]benzoic acid methyl ester 20 (21.5 g, 38.5 mmol) was hydrolyzed as described above for the preparation of Example 2. The acid was converted to the sodium salt and purified as described above for the preparation of Example 1(D) to provide 16.7 g (77%) of the desired title product as a white amorphous solid: NMR 25  $(DMSO-d_6)$  10.50 (bs, 1H, -OH), 7.51 (m, 3H), 7.20 (t, J = 7.4 Hz, 1H), 7.13 (m, 2H), 7.00 (m, 2H), 6.95 (s, 1H), 6.67 (dd, J = 8.2, 3.3 Hz, 2H), 6.62 (s, 1H), 6.26 (d, J = 8.2)Hz, 1H), 4.14 (t, J = 5.8 Hz, 2H), 4.02 (t, J = 5.7 Hz, 2H), 2.60 (t, J = 6.8 Hz, 2H), 2.47 (q, J = 7.3 Hz, 2H), 2.16 (t, J = 6.8 Hz, 2H), 2.16 (t, J = 6.830  $J = 5.9 \text{ Hz}, 2H), 1.45 \text{ (hextet, } J = 7.5 \text{ Hz}, 2H), 1.07 \text{ (t, } J = 7.5 \text$ 7.5 Hz, 3H), 0.81 (t, J = 7.4 Hz, 3H); MS-FAB m/e 568 (38, p + 1), 567 (100, p), 544 (86), 527 (77), 295 (65), 253 (45); IR  $(KBr, cm^{-1})$  3407 (b), 2962, 1603, 1502, 1446, 1395, 1239, 1112. 35

Analysis for C33H32O6FNa:

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Calc: C, 69.95; H, 5.69; F, 3.35;

Found: C, 69.97; H, 5.99; F, 3.52.

The methods of the present invention describe the use of leukotriene antagonists for the inhibition and treatment of ischemia reperfusion injury which is characterized by the excessive release of leukotriene B4.

The term "excessive release" of a leukotriene refers to an amount of the leukotriene sufficient to cause ischemia reperfusion injury. The amount of leukotriene which is considered to be excessive will depend on a variety of factors, including the amount of leukotriene required to cause the disease, and the species of the mammal involved. As will be appreciated by those skilled in the art, the success of treating a mammal suffering from ischemia reperfusion injury by an excessive release of leukotriene with a compound of Formula I will be measured by the regression or prevention of the symptoms of the condition.

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#### <u>Assays</u>

#### Assay 1

The effectiveness of compounds of Formula I to inhibit the binding of tritiated LTB4 to guinea pig lung membranes was determined as follows.

## [3H]-LTB4 Radioligand Binding Assay in Guinea Pig Lung Membranes

[3H]-LTB4 (196-200 Ci/mmole) was purchased from New England Nuclear (Boston, MA). All other materials were purchased from Sigma (St. Louis, MO). Incubations (555 mL) were performed in polypropylene minitubes for 45 minutes at 30°C and contained 25 mg of guinea pig lung membrane protein (Silbaugh, et al., European Journal of Pharmacology, 223 (1992) 57-64) in a buffer containing 25 mM MOPS, 10 mM

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MgCl<sub>2</sub>, 10 mM CaCl<sub>2</sub>, pH 6.5, approximately 140 pM [<sup>3</sup>H]-LTB<sub>4</sub>, and displacing ligand or vehicle (0.1% DMSO in 1 mM sodium carbonate, final concentration) as appropriate. The binding reaction was terminated by the addition of 1 mL ice cold 5 wash buffer (25 mM Tris-HCl, pH 7.5) followed immediately by vacuum filtration over Whatman GF/C glass fiber filters using a Brandel (Gaithersburg, MD) 48 place harvester. The filters were washed three times with 1 mL of wash buffer. Retained radioactivity was determined by liquid scintillation counting at 50% counting efficiency using 10 Ready Protein Plus cocktail (Beckman, Fullerton, CA). Nondisplaceable binding was determined in the presence of 1 mM LTB4 and was usually less than 10% of total binding. Data were analyzed using linear regression analysis of log-logit plots of the values between 10% and 90% of control binding 15 to calculate IC50s and slope factors (pseudo-Hill coefficients). IC50 values thus obtained were corrected for radioligand concentration (Cheng and Prusoff, Biochem. Pharmacol., 22, 3099 (1973)) to calculate Ki values. pKi is the mean -log Ki for n experiments. 20

Compounds of the instant invention tested in the above assay were found to have a pKi of between 7 and 10.

The ability of a compound of formula I to ameliorate tissue damage caused by an ischemia-reperfusion insult in different tissues can be demonstrated in a variety of animal species. The methodology for showing protective action in splanchnic ischemia/reperfusion injury in rats is an example (Karasawa et al., Am. J. Physiol., 261, G191-8,1991).

Assay 2

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In this circulatory shock model, ischemia is induced by clamping both the celiac and superior mesenteric arteries with atraumatic arterial clamps. Immediately after occlusion, rats are given heparin (500 U/kg i.v.). The clamps are removed after 90 minutes and splanchnic vascular

reperfusion allowed to proceed. Animals are followed for an additional 2 hours or until the mean arterial blood pressure (MABP) falls below 45 mmHg. Survival time is the period from removal of the clamps to the time MABP falls below 45 mmHg. Survivors are defined as animals that maintain a MABP of 45 mmHg for 120 min after reperfusion. A dose response is obtained by dividing the animals into 5 experimental groups of 10 rats each. The groups are: sham-treated rats given vehicle: occluded-reperfused (O/R) rats given vehicle; O/R rats given 1, 3, or 10 mg/kg of compound intravenously 10 10 minutes before reperfusion. At the end of the experiment, blood is withdrawn for analysis of tissue breakdown products such as plasma free amino-nitrogen and myocardial depressant factor (MDF). Also, a 5-cm piece of ileal tissue is excised about 30 cm distal from the stomach 15 and analyzed for myeloperoxidase activity, a marker for neutrophil infiltration. The effectiveness of a treatment is accessed by comparing survival time, percent survivors, free amino-nitrogen concentration, MDF concentration and ileum MPO activity of the treated group to that of a vehicle 20 control.

The therapeutic and prophylactic treatments provided by this invention are practiced by administering to a mammal in need thereof a dose of a compound of formula I or a pharmaceutically acceptable salt or solvate thereof, that is effective to inhibit or treat ischemia reperfusion injury.

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The term "inhibit" includes its generally accepted meaning which includes prohibiting, preventing, restraining and slowing, stopping or reversing progression, severity or a resultant symptom. As such, the present method includes both medical therapeutic and/or prophylactic administration as appropriate.

While it is possible to administer a compound employed in the methods of this invention directly without any formulation, the compounds are usually administered in the form of pharmaceutical formulation comprising a pharmaceutically acceptable excipient and at least one

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compound of the present invention. The compounds or formulations of the present invention may be administered by the oral and rectal routes, topically, parenterally, e.g., by injection and by continuous or discontinuous intraarterial infusion, in the form of, for example, tablets, lozenges, sublingual tablets, sachets, cachets, elixirs, gels, suspensions, aerosols, ointments, for example, containing from 0.01 to 90% by weight of the active compound in a suitable base, soft and hard gelatin capsules, suppositories, injectable solutions and suspensions in 10 physiologically acceptable media, and sterile packaged powders adsorbed onto a support material for making injectable solutions. Such formulations are prepared in a manner well known in the pharmaceutical art and comprise at least one active compound. See, e.g., REMINGTON'S 15 PHARMACEUTICAL SCIENCES, (16th ed. 1980).

In making the formulations employed in the present invention the active ingredient is usually mixed with an excipient, diluted by an excipient or enclosed within such a carrier which can be in the form of a capsule, sachet, paper or other container. When the excipient serves as a diluent, it can be a solid, semi-solid, or liquid material, which acts as a vehicle, carrier or medium for the active ingredient.

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In preparing a formulation, it may be necessary to mill the active compound to provide the appropriate particle size prior to combining with the other ingredients. If the active compound is substantially insoluble, it ordinarily is milled to a particle size of less than 200 mesh. If the active compound is substantially water soluble, the particle size is normally adjusted by milling to provide a substantially uniform distribution in the formulation, e.g. about 40 mesh.

Some examples of suitable carriers, excipients and diluents include lactose, dextrose, sucrose, sorbitol, mannitol, starches, gum acacia, calcium phosphate, alginates, tragacanth, gelatin, calcium silicate,

microcrystalline cellulose, polyvinylpyrrolidone, cellulose, tragacanth, gelatin, water, syrup, and methyl cellulose. The formulations can additionally include: lubricating agents such as talc, magnesium stearate, and mineral oil; wetting agents; emulsifying and suspending agents; preserving agents such as methyl- and propylhydroxybenzoates; sweetening agents; and flavoring agents. The compositions of the invention can be formulated so as to provide quick, sustained or delayed release of the active ingredient after administration to the patient by employing procedures known in the art.

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The compounds of this invention may be delivered transdermally using known transdermal delivery systems and excipients. Most preferably, a compound of this invention is admixed with permeation enhancers including, but not limited to, propylene glycol, polyethylene glycol monolaurate, and azacycloalkan-2-ones, and incorporated into a patch or similar delivery system. Additional excipients including gelling agents, emulsifiers, and buffers may be added to the transdermal formulation as desired.

For topical administration, a compound of this invention ideally can be admixed with any variety of excipients in order to form a viscous liquid or cream-like preparation.

For oral administration, a compound of this invention ideally can be admixed with carriers and diluents and molded into tablets or enclosed in gelatin capsules.

In the case of tablets, a lubricant may be incorporated to prevent sticking and binding of the powdered ingredients in the dies and on the punch of the tableting machine. For such purpose there may be employed for instance aluminum, magnesium or calcium stearates, talc or mineral oil.

Preferred pharmaceutical forms of the present invention parenteral, capsules and tablets.

The therapeutic and prophylactic treatments provided by this invention are practiced by administering to a mammal in need thereof a dose of a compound of formula I or a

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pharmaceutically acceptable salt or solvate thereof that is effective to inhibit or treat ischemia reperfusion injury.

Advantageously for this purpose, formulations may be provided in unit dosage form, preferably each dosage unit containing from about 5 to about 500 mg (from about 5 to 50 mg in the case of parenteral or inhalation administration, and from about 25 to 500 mg in the case of oral or rectal administration) of a compound of Formula I. Dosages from about 0.5 to about 300 mg/kg per day, preferably 0.5 to 20 mg/kg, of active ingredient may be administered although it will, of course, readily be understood that the amount of the compound or compounds of Formula I actually to be administered will be determined by a physician, in the light of all the relevant circumstances including the condition to be treated, the choice of compound to be administered and the choice of route of administration and therefore the above preferred dosage range is not intended to limit the scope of the present invention in any way.

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The specific dose of a compound administered according to this invention to obtain therapeutic or prophylactic effects will, of course, be determined by the particular circumstances surrounding the case, including, for example, the route of administration the age, weight and response of the individual patient, the condition being treated and the severity of the patient's symptoms.

In general, the compounds of the invention are most desirably administered at a concentration that will generally afford effective results without causing any serious side effects and can be administered either as a single unit dose, or if desired, the dosage may be divided into convenient subunits administered at suitable times throughout the day.

While all of the compounds illustrated above exemplify LTB4 inhibition activity in vitro, we have also discovered that compounds bearing a single acidic group  $(R_6)$  are considerably more orally bioactive when administered to mammals compared with those compounds bearing two such

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acidic groups. Thus, a preferred embodiment when administering compounds of Formula I orally to mammals comprises administering compounds bearing a single acidic  $R_6$  functionality.

The following formulation examples may employ as active compounds any of the compounds of this invention. The examples are illustrative only and are not intended to limit the scope of the invention in any way.

10 Formulation 1

Hard gelatin capsules are prepared using the following ingredients:

Ouantity (mg/capsule)

3-(2-(3-(2-Ethyl-4-(4-fluorophenyl)-5hydroxyphenoxy)propoxy)-6-(4-carboxyphenoxy)phenyl)propanoic acid 250 Starch 200 Magnesium stearate 10

The above ingredients are mixed and filled into hard gelatin capsules in 460 mg quantities.

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#### Formulation 2

A tablet is prepared using the ingredients below:

5	<u>Ouanti</u>	ty (mg/tablet)
	1-(4-(Carboxymethoxy)phenyl)-1-(1H-	•
•	tetrazol-5-yl)-6-(2-ethyl-4-(4-	
	fluorophenyl)-5-hydroxyphenoxy)hexane	250
10	Cellulose, microcrystalline	400
	Silicon dioxide, fumed	10
	Magnesium stearate	5

The components are blended and compressed to form tablets each weighing 665 mg.

#### Formulation 3

An aerosol solution is prepared containing the 20 following components:

	Weight *				
3-[4-[7-Carboxy-9-oxo-3-[3-[2-ethyl-	-4-				
(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]-					
9H-xanthene]]propanoic acid	0.25				
Ethanol	30.00				
Propellant 11	10.25				
(trichlorofluoromethane)					
Propellant 12	29.75				
(Dichlorodifluoromethane)					
Propellant 114	29.75				
(Dichlorotetrafluoroethane)					
	(4-fluorophenyl)-5-hydroxyphenoxy 9H-xanthene]]propanoic acid Ethanol Propellant 11     (trichlorofluoromethane) Propellant 12     (Dichlorodifluoromethane) Propellant 114				

35 The active compound is dissolved in the ethanol and the solution is added to the propellant 11, cooled to  $-30\,^{\circ}$ C. and transferred to a filling device. The required amount is then

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fed to a container and further filled with the pre-mixed propellants 12 and 114 by means of the cold-filled method or pressure-filled method. The valve units are then fitted to the container.

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#### Formulation 4

Tablets each containing 60 mg of active ingredient are made up as follows:

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	2-[2-Propyl-3-[3-[2-ethyl-5-hydroxy-4-(4-	
	fluorophenyl)phenoxy]propoxy]phenoxy]-	
	benzoic acid sodium salt	60 mg
15	Starch	45 mg
	Microcrystalline cellulose	35 mg
	Polyvinylpyrrolidone	4 mg
	(as 10% solution in water)	
	Sodium carboxymethyl starch	4.5 mg
20	Magnesium stearate	0.5 mg
	Talc	1 mg
	Total	150 mg

The active ingredient, starch and cellulose are passed through a No. 45 mesh U.S. sieve and mixed thoroughly. The solution of polyvinylpyrrolidone is mixed with the resultant powders which are then passed through a No. 14 mesh U.S. sieve. The granules so produced are dried at 50-60° and passed through a No. 18 mesh U.S. sieve. The sodium carboxymethyl starch, magnesium stearate and talc, previously passed through a No. 60 mesh U.S. sieve, are then added to the granules which, after mixing, are compressed on a tablet machine to yield tablets each weighing 150 mg.

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#### Formulation 5

Capsules each containing 80 mg of medicament are made as follows:

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	5-[3-[2-(1-Carboxy)ethyl]-4-[3-[2-ethyl-4-(				
	fluorophenyl)-5-hydroxyphenoxy]	propoxy]-			
	phenyl]-4-pentynoic acid	80 mg			
	Starch	59 mg			
10	Microcrystalline cellulose	59 mg			
	Magnesium stearate	2 mg			
	Total	200 mg			

The active ingredient, cellulose, starch and magnesium stearate are blended, passed through a No. 45 mesh U.S. sieve, and filled into hard gelatin capsules in 200 mg quantities.

#### Formulation 6

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Suppositories each containing 225 mg of active ingredient are made as follows:

The active ingredient is passed through a No. 60 mesh U.S. sieve and suspended in the fatty acid glycerides previously melted using the minimum heat necessary. The mixture is then poured into a suppository mold of nominal 2 g capacity and allowed to cool.

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#### Formulation 7

Suspensions each containing 50 mg of medicament per 5 mL dose are made as follows:

5		5-hydroxyphenoxy]propoxy]phenoxy]benzoic acid 50 mg dium carboxymethyl cellulose 50 mg gar 1 g thyl paraben 0.05 mg opyl paraben 0.03 mg avor q.v.		
	2-[2-Propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-			
	5-hydroxyphenoxy]propoxy]phenoxy]	benzoic		
	acid	50 mg		
	Sodium carboxymethyl cellulose	50 mg		
10	Sugar	1 g		
	Methyl paraben	0.05 mg		
	Propyl paraben	0.03 mg		
	Flavor	q.v.		
	Color	q.v.		

Purified water to

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The medicament is passed through a No. 45 mesh U.S. sieve and mixed with the sodium carboxymethylcellulose, sugar, and a portion of the water to form a suspension. The parabens, flavor and color are dissolved and diluted with some of the water and added, with stirring. Sufficient water is then added to produce the required volume.

5 mL

#### Formulation 8

25 An intravenous formulation may be prepared as follows:

The solution of the above ingredients generally is

30 administered intravenously to a subject at a rate of 1 ml
per minute.

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We claim:

A method for inhibiting or treating ischemia
 reperfusion injury in a mammal which comprises administering to a mammal in need thereof an effective amount of a compound of the formula I

$$R_3$$
 $R_2$ 
 $R_2$ 
 $R_3$ 
 $R_1$ 

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wherein:

R<sub>1</sub> is C<sub>1</sub>-C<sub>5</sub> alkyl, C<sub>2</sub>-C<sub>5</sub> alkenyl, C<sub>2</sub>-C<sub>5</sub> alkynyl,

C<sub>1</sub>-C<sub>4</sub> alkoxy, (C<sub>1</sub>-C<sub>4</sub> alkyl)thio, halo, or R<sub>2</sub>
substituted phenyl;

each R<sub>2</sub> and R<sub>3</sub> are each independently hydrogen, halo, hydroxy, C<sub>1</sub>-C<sub>4</sub> alkyl, C<sub>1</sub>-C<sub>4</sub> alkoxy, (C<sub>1</sub>-C<sub>4</sub> alkyl)-S(O)<sub>q</sub>-, trifluoromethyl, or di-(C<sub>1</sub>-C<sub>3</sub> alkyl)amino;

 $X \text{ is } -O-, -S-, -C(=O), \text{ or } -CH_2-;$ 

25 Y is -O- or -CH<sub>2</sub>-;

or when taken together, -X-Y- is -CH=CH- or

—<u>c</u>=c— ;

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Z is a straight or branched chain  $C_1$ - $C_{10}$  alkylidenyl;

A is a bond, -O-, -S-, -CH=CH-, or -CRaRb-, where Ra and Rb are each independently hydrogen, C1-C5 alkyl, or R7-substituted phenyl, or when taken together with the carbon atom to which they are attached form a C4-C8 cycloalkyl ring;

R4 is R6

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$$R_{11}$$

$$W-R_6$$
 or  $W-R_6$ 

where,

5

each R6 is independently -COOH, 5-tetrazolyl, -CON(R9)2, or -CONHSO $_2$ R10;

10

each R7 is hydrogen, C1-C4 alkyl, C2-C5 alkenyl, C2-C5 alkynyl, benzyl, methoxy, -W-R6, -T-G-R6, (C1-C4 alkyl)-T-(C1-C4 alkylidenyl)-O-, or hydroxy;

Rg is hydrogen or halo;

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each R9 is independently hydrogen, phenyl, or C1-C4 alkyl, or when taken together with the

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nitrogen atom form a morpholino, piperidino, piperazino, or pyrrolidino group;

R10 is C1-C4 alkyl or phenyl;

5

R<sub>11</sub> is R<sub>2</sub>, -W-R<sub>6</sub>, or -T-G-R<sub>6</sub>;

each W is a bond or straight or branched chain divalent hydrocarbyl radical of one to eight

10 carbon atoms;

each G is a straight or branched chain divalent hydrocarbyl radical of one to eight carbon atoms;

each T is a bond,  $-CH_2-$ , -O-, -NH-, -NHCO-, -C(=O)-, or  $-S(O)_q-$ ;

K is -C(=O) - or -CH(OH) -;

each q is independently 0, 1, or 2;

p is 0 or 1; and

t is 0 or 1;

25

provided when X is -O- or -S-, Y is not -O-;

provided when A is -O- or -S-, R4 is not R6;

provided when A is -O- or -S- and Z is a bond, Y is not -O-; and

provided W is not a bond when p is 0;

35 or a pharmaceutically acceptable salt or solvate thereof.

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2. The method as claimed in **Claim 1** employing a compound of the formula

$$R_2$$
 O-CH<sub>2</sub>-Z-A-R<sub>4</sub>

or a pharmaceutically acceptable salt or solvate thereof.

- 3. The method as claimed in **Claim 2** employing 2-[2-propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]phenoxy]benzoic acid or a pharmaceutically acceptable salt or solvate thereof.
- 4. The method as claimed in Claim 2 employing 3-(2-(3-(2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy)propoxy) 6-(4-carboxy-phenoxy)phenyl)propionic acid or a pharmaceutically acceptable salt or solvate thereof.
- The method as claimed in Claim 2 employing 1-(4-(carboxy-methoxy)phenyl)-1-(1H-tetrazol-5-yl)-6-(2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy)hexane or a pharmaceutically acceptable salt or solvate thereof.
- 6. The method as claimed in **Claim 2** employing 3-[4-[7-carboxy-9-oxo-3-[3-[2-ethyl-4-(4-fluorophenyl)-5hydroxyphenoxy]-propoxy]-9H-xanthene]]propanoic acid or a pharmaceutically acceptable salt or solvate thereof.
- 7. The method as claimed in **Claim 2** employing 5[3-[2-(1-carboxy)-ethyl]-4-[3-[2-ethyl-4-(4-fluorophenyl)-530 hydroxyphenoxy]-propoxy]phenyl]-4-pentynoic acid or a
  pharmaceutically acceptable salt or solvate thereof.

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8. The method as claimed in any one of Claims 1 to 7 in which the mammal is a human.

9. Use of a compound of the formula I

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$$R_{2}$$
 $R_{2}$ 
 $R_{3}$ 
 $R_{1}$ 

wherein:

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R1 is C1-C5 alkyl, C2-C5 alkenyl, C2-C5 alkynyl, C1-C4 alkoxy, (C1-C4 alkyl)thio, halo, or R2-substituted phenyl;

15

each R2 and R3 are each independently hydrogen, halo, hydroxy, C1-C4 alkyl, C1-C4 alkoxy, (C1-C4 alkyl)-S(0) $_{\bf q}$ -, trifluoromethyl, or di-(C1-C3 alkyl)amino;

20

 $X \text{ is } -O-, -S-, -C(=O), \text{ or } -CH_2-;$ 

Y is -O- or -CH2-;

or when taken together, -X-Y- is -CH=CH- or

25

Z is a straight or branched chain  $C_1$ - $C_{10}$  alkylidenyl;

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A is a bond, -O-, -S-, -CH=CH-, or -CRaRb-, where  $R_{a}$  and  $R_{b}$  are each independently hydrogen, C1-C5 alkyl, or R7-substituted phenyl, or when taken together with the carbon atom to which they are attached form a C4-C8 cycloalkyl ring;

 $R_4$  is  $R_6$ 

$$R_{11}$$

$$W^{-R_6}$$
 or  $W^{-R_6}$ 

where,

5

each R6 is independently -COOH, 5-tetrazolyl,
-CON(R9)2, or -CONHSO2R10;

10

each R7 is hydrogen, C1-C4 alkyl, C2-C5 alkenyl, C2-C5 alkynyl, benzyl, methoxy, -W-R6, -T-G-R6, (C1-C4 alkyl)-T-(C1-C4 alkylidenyl)-O-, or hydroxy;

R8 is hydrogen or halo;

15

each R9 is independently hydrogen, phenyl, or  $C_1$ - $C_4$  alkyl, or when taken together with the

nitrogen atom form a morpholino, piperidino, piperazino, or pyrrolidino group;

R10 is C1-C4 alkyl or phenyl;

5

R11 is R2, -W-R6, or -T-G-R6;

each W is a bond or straight or branched chain divalent hydrocarbyl radical of one to eight carbon atoms;

10

each G is a straight or branched chain divalent hydrocarbyl radical of one to eight carbon atoms;

15

each T is a bond, -CH<sub>2</sub>-, -O-, -NH-, -NHCO-, -C(=O)-, or -S(O)<sub>Q</sub>-;

K is -C(=0) - or -CH(OH) -;

20

each q is independently 0, 1, or 2;

p is 0 or 1; and

t is 0 or 1;

25

provided when X is -O- or -S-, Y is not -O-;

provided when A is -O- or -S-, R4 is not R6;

30

provided when A is -O- or -S- and Z is a bond, Y is not -O-; and

provided W is not a bond when p is 0;

or a pharmaceutically acceptable salt or solvate thereof, optionally in combination with a pharmaceutically acceptable excipient, for the preparation of a pharmaceutical

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composition for inhibiting or treating ischemia reperfusion injury in a mammal.

10. The use according to claim 9 employing a5 compound of the formula;

$$R_2$$
 $O-CH_2-Z-A-R_4$ 
 $R_1$ 

or a pharmaceutically acceptable salt or solvate thereof.

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11. The use according to claim 9 wherein the compound employed is 2-[2-propyl-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]propoxy]phenoxy]benzoic acid or a pharmaceutically acceptable salt or solvate thereof.

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12. The use according to claim 9 wherein the compound employed is 3-(2-(3-(2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy)propoxy)-6-(4-carboxy-phenoxy)phenyl)propionic acid or a pharmaceutically acceptable salt or solvate thereof.

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13. The use according to claim 9 wherein the compound employed is 1-(4-(carboxy-methoxy)phenyl)-1-(1H-tetrazol-5-yl)-6-(2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy)hexane or a pharmaceutically acceptable salt or solvate thereof.

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14. The use according to claim 9 wherein the compound employed is 3-[4-[7-carboxy-9-oxo-3-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]-propoxy]-9H-xanthene]]propanoic acid or a pharmaceutically acceptable salt or solvate thereof.

15. The use according to claim 9 wherein the compound employed is 5-[3-[2-(1-carboxy)-ethyl]-4-[3-[2-ethyl-4-(4-fluorophenyl)-5-hydroxyphenoxy]-propoxy]phenyl]-5 4-pentynoic acid or a pharmaceutically acceptable salt or solvate thereof.

#### INTERNATIONAL SEARCH REPORT

International application No. PCT/US98/05478

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US CL	:A61K 31/52, 31/41, 31/19 :514/263, 381, 568, 570				
	to International Patent Classification (IPC) or to both	national classification and IPC			
	DS SEARCHED	d by classification symbols			
	ocumentation searched (classification system follows	d by classification symbols;			
U.S. :	514/263, 381, 568, 570				
Documenta	tion searched other than minimum documentation to the	extent that such documents are included	in the fields searched		
Electronic o	data base consulted during the international search (na	ame of data base and, where practicable	e, search terms used)		
C. DOC	UMENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where ap	propriate, of the relevant passages	Relevant to claim No.		
x	US 5,462,954 A (BAKER et al.) 31 C		9-15		
 Y	line 39 - column 3, line 20 and column line 24.	1 135, line 33 - column 137,	1-8		
1	inc 24.				
Y	US 5,492,915 A (DEREU et al.) 20 Folines 59-67.	ebruary 1996, see column 1,	1-8		
Y	US 5,457,124 A (COHEN et al.) 10 October 1995, see column 1, 1-8 lines 50-60.				
Furd	her documents are listed in the continuation of Box C	. See patent family annex.			
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